

PROBLEM 2.52

KNOWN: Data are provided for a grill hood.

FIND: Determine the net rate of heat transfer between the hood and the surroundings by convection and radiation, per unit area of hood surface.

SCHEMATIC & GIVEN DATA:

$$T_0 = 27^\circ\text{C}$$

$$h = 10 \text{ W/m}^2 \cdot \text{K}$$

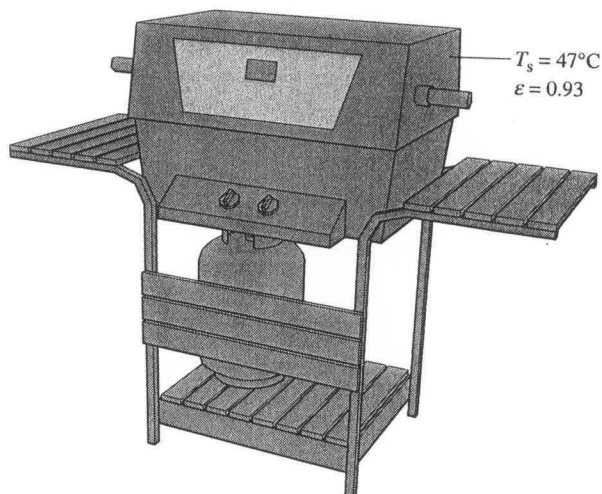


Fig. P2.52

ENGR. MODEL:

1. Radiative heat transfer between the hood and surroundings is modeled as an exchange between a surface at T_s and a much larger surrounding surface at T_0 .

ANALYSIS:

Using Eq. 2.34, convection heat transfer is

$$\frac{\dot{Q}_c}{A} = h[T_s - T_0] = 10 \frac{\text{W}}{\text{m}^2 \cdot \text{K}} [20 \text{ K}] \left| \frac{1 \text{ kW}}{10^3 \text{ W}} \right|$$

$$= 0.2 \text{ kW/m}^2$$

Using Eq. 2.33, radiative heat transfer is

$$\frac{\dot{Q}_e}{A} = \epsilon \sigma [T_s^4 - T_0^4]$$

$$= 0.93 [5.67 \times 10^{-8} \frac{\text{W}}{\text{m}^2 \cdot \text{K}^4}] [(320 \text{ K})^4 - (300 \text{ K})^4]$$

$$= 125.8 \frac{\text{W}}{\text{m}^2} \left| \frac{1 \text{ kW}}{10^3 \text{ W}} \right| = 0.13 \text{ kW/m}^2$$

$$\text{Total} = (\dot{Q}_c/A) + (\dot{Q}_e/A) = 0.33 \text{ kW/m}^2$$

PROBLEM 2.53

KNOWN: Data is provided for $Q, W, E_1, E_2, \Delta E$ for several processes of a closed system.

FIND: Determine values of missing table data.

Process	Q	W	E_1	E_2	ΔE
a	+50	-20	-20	+50	+70
b	+50	+20	+20	+50	+30
c	-40	-60	+40	+60	+20
d	-90	-90	+50	+50	0
e	+50	+150	+20	-80	-100

$$\left. \begin{array}{l} \text{a} \\ \text{b} \\ \text{c} \\ \text{d} \\ \text{e} \end{array} \right\} \Delta E = Q - W$$

Process a:

$$\Delta E = Q - W$$

$$\therefore Q = \Delta E + W = 70 + (-20) = \boxed{+50}$$

$$\Delta E = E_2 - E_1 = +70 \Rightarrow E_1 = E_2 - 70$$

$$E_1 = 50 - 70 = \boxed{-20}$$

Process b: $\Delta E = E_2 - E_1 = 50 - 20 = \boxed{30}$

$$\Delta E = Q - W$$

$$\therefore W = Q - \Delta E = +50 - 30 = \boxed{20}$$

Process c:

$$\Delta E = E_2 - E_1 \Rightarrow E_1 = E_2 - \Delta E$$

$$= 60 - 20 = \boxed{40}$$

$$\Delta E = Q - W \Rightarrow Q = \Delta E + W$$

$$\therefore Q = 20 + (-60) = \boxed{-40}$$

Process d:

$$\Delta E = E_2 - E_1 \Rightarrow E_1 = E_2 - \Delta E$$

$$\therefore E_1 = 50 - 0 = \boxed{50}$$

$$\Delta E = Q - W \Rightarrow Q = \Delta E + W$$

$$\therefore Q = 0 + (-90) = \boxed{-90}$$

Process e:

$$\Delta E = Q - W = 50 - 150 = \boxed{-100}$$

$$\Delta E = E_2 - E_1 \Rightarrow E_2 = \Delta E + E_1$$

$$E_2 = -100 + 20 = \boxed{-80}$$