

INSTRUCTOR GUIDE

BIOLOGY *Science for Life*

and

BIOLOGY *Science for Life*

WITH PHYSIOLOGY

FOURTH EDITION

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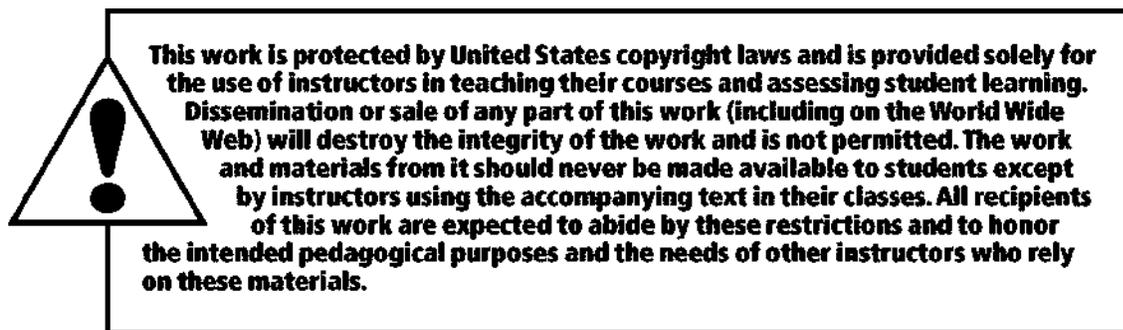
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Can Science Cure the Common Cold?

Introduction to the Scientific Method

OBJECTIVES

Teaching Goals

Most students have covered the scientific method in high school biology. However, when asked to define a hypothesis, the majority will say, “A hypothesis is an educated guess,” without any comprehension of what it really means. This chapter introduces the student to the scientific method as a way of thinking and explaining observations. This chapter also familiarizes students in the proper setup of experiments and statistical analysis of data. The emphasis should be on using critical thinking in everyday decision making—not just in a laboratory. Students are inundated with claims based on anecdotal evidence and misrepresentation of data. Your goal is to help the students evaluate such claims by critical analysis of the data.

Student Goals

By the end of this chapter, students should be able to accomplish the following learning objectives:

- Describe the characteristics of a scientific hypothesis.
- Compare and contrast the terms *scientific hypothesis* and *scientific theory*.
- Distinguish between inductive and deductive reasoning.
- Explain why the truth of a hypothesis cannot be proven conclusively via deductive reasoning.
- Describe the features of a controlled experiment and explain how these experiment eliminate alternative hypotheses for the results.
- List strategies for minimizing bias when designing experiments.
- Define correlation, and explain the benefits and limitations of using this technique to test hypotheses.
- Describe the information that statistical tests prove.
- Compare and contrast primary and secondary sources.
- Summarize the techniques you can use to evaluate scientific information from secondary sources.

LECTURE OUTLINE

1.1 The Process of Science (Figure 1.1)

- A. Scientific method allows for solving of problems and answering questions.
- B. Hypotheses are explanations of how something works and must be testable and falsifiable.

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- C. Scientific theories are supported by multiple lines of evidence from hypotheses that have been tested by deductive reasoning.
- D. Construction of hypothesis involves inductive reasoning and hypothesis testing involves deductive reasoning. **(Figure 1.3)**

1.2 Hypothesis Testing

- A. Hypotheses are tested by experiments that should be controlled for variables and bias.
- B. Correlation between variables may provide answers when other testing is problematic. **(Figure 1.10, Figure 1.11)**
- C. Model systems are used when testing on humans is dangerous or unethical. **(Figure 1.7)**

Lecture Activity 1.1

Lecture Activity 1.2

Lecture Activity 1.3

1.3 Understanding Statistics

- A. Statistical analysis can help researchers apply results from a small experimental group to a larger group without actually testing the larger group. **(Figure 1.12)**
- B. The significance of results is influenced by the size of the sample and the validity of the hypothesis. **(Figure 1.15)**
- C. Sampling error is the difference between a sample and the population that it is being compared to.

1.4 Evaluating Scientific Information

- A. Information from primary sources is used as the basis for news articles and websites. **(Figure 1.16)**
- B. Different media sources have various levels of credibility and reliability. **Lecture Activity 1.4**

1.5 Is There a Cure for the Common Cold?

- A. Hand-washing is the best method for preventing the common cold.

Lecture Activity 1.1: Developing and Testing Hypotheses

Estimated Time to Complete: 15–20 minutes, or longer with discussion

Introduction: This activity gives students a chance to practice what they have learned about developing hypotheses and testing them. Students work in groups of three to four to promote interactive learning and discussion. An observation is given to each group, and the group members' job is to create a testable and falsifiable hypothesis to explain that observation. The activity can stand alone, or it can become the basis of a discussion after completion. If desired, groups could report to the class on their observation, hypothesis, and experiment.

A handout is provided with a sample observation. Here are several more suggestions, and others can be made up as needed:

- Algae appear in clean water left outside in a child's plastic pool for a week.
- Lawn grass doesn't grow at the beach.
- Mosquitoes bite some people more than others.
- Coleus plants grown inside are taller but have thinner stems than coleus plants grown outside.
- Ants eat some foods that they find outside, but not all.
- Bees visit some types of flowers, but not others.

Material

- Lecture Activity 1.1 Handout: Developing and Testing Hypotheses

Procedures

1. Divide the students into groups.
2. Clearly explain that each group will receive an observation and that their job is first to formulate a testable hypothesis to explain that observation. They will then come up with an experimental protocol, including controls that would test their hypothesis.
3. Distribute a handout to each group, or give them their observation verbally. Groups can all receive the same observation, or a different observation can be used for each group.
4. Give the students 10–15 minutes to generate their hypothesis and experiment. Circulate around the classroom during that time to make sure that groups remain on task, that their hypotheses are reasonable and testable, and that the experimental procedure includes a control and makes sense.
5. If desired, ask groups to present their observation, hypothesis, and experiment to the class.

Assessment Suggestions: Have groups turn in their papers. Check that the hypotheses are testable and falsifiable and that control groups were used properly. Give students participation points for presenting their ideas to the class.

For an online course have students submit their hypotheses in a discussion forum and have the students discuss the results and determine if each hypothesis is in the correct format. Points are given following a rubric for participation.

Lecture Activity 1.2: Do Bees See in Color?

Estimated Time to Complete: 15–20 minutes, or longer with discussion

Introduction: This activity puts students into the shoes of a real scientist who is trying to answer a question about nature. The students read an account of some actual experiments done by Karl von Frisch to try to determine whether bees can see in color. They start with von Frisch’s observations, find out what he did, and work through the experimental method along with him. Working in groups, they see how von Frisch interpreted and then improved his experimental method to draw his conclusions.

Material

- Lecture Activity 1.2 Handout: Do Bees See in Color?

Procedures

1. Divide students into groups.
2. Briefly introduce the activity and explain what students are to do.
3. Pass out handouts.
4. As students are working, circulate around the classroom to check that groups remain on track. Sometimes they can get off on a tangent, hypothesizing, for example, that bees like sugar water. They also tend to express themselves in vague generalities; keep them to specifics.

Assessment Suggestions: Collect and check group papers. A key for correcting the handout is provided. Tell students that, using a similar set of experiments, von Frisch demonstrated that bees could not see the color red. Ask the class to describe the experiments and results that would have led von Frisch to reach this conclusion.

You could also put the handout online and have them fill out the handout online after completing the work in the classroom or have students work on this individually online and treat it like an assignment.

Handout Answer Key

1. What was von Frisch’s hypothesis? Von Frisch hypothesized that bees could see the color blue.
2. What conclusion should von Frisch draw from his results? He should conclude that the results supported his hypothesis—that the bees could distinguish the blue paper from the red paper.
3. What observation would have falsified von Frisch’s hypothesis? His hypothesis would have been falsified if the bees had swarmed around both bowls equally.
4. Which uncontrolled variable in his first experiment did von Frisch correct in the second experiment, and why was it important? The uncontrolled variable was color intensity. It is important because it is a way that bees could distinguish between the papers without actually being able to see their color.
5. How did von Frisch control for this variable in the second experiment? He put bowls on a series of gray papers with similar color intensity to the blue paper to see if that would confuse the bees.
6. What was the overall conclusion from von Frisch’s series of experiments? (Remember to refer to his original hypothesis.) Von Frisch concluded that his hypothesis was supported; bees can see the color blue.

Lecture Activity 1.3: Demonstrating the Importance of Control Groups

Estimated Time to Complete: 10–15 minutes

Introduction: This short demonstration helps students understand the importance of a control group in drawing inferences from data. Students are presented with data on the effect of drugs on the survival rates of several diseases, but then will realize that the data are meaningless without the ability to compare the experimental data to a control group.

Material

- Copy or project the data given in steps 2 and 4.

Procedures

1. Ask students to imagine that they head up a research expedition to a remote location; they have a medical computer and some medicines, but no doctor with them. Three members of the research team come down with different illnesses, and the expedition leader must decide how to treat them. The medical computer diagnoses their illnesses and finds out some basic data on treatment options. Luckily, some of the drugs that have been used to treat their conditions in the past are available. The computer finds the information given in step 2.
2. Copy the following data onto a slide or transparency, or write it on the board:
Disease A, treated with drug A, has a survival rate of 23%.
Disease B, treated with drug B, has a survival rate of 56%.
Disease C, treated with drug C, has a survival rate of 89%.

Based on these data, ask students what they think of these treatments and how they would choose to treat their assistants.

3. On giving the matter further thought, you realize that you are missing some important data. You can't really understand the efficacy of the drug treatment unless you know the survival rates in the control groups that received no medications. A little more digging gives you additional data (see step 4).
4. Deliver the following data to the students in the same way, so that they can see both sets of data at the same time:
Disease A, untreated, has a survival rate of 10%.
Disease B, untreated, has a survival rate of 54%.
Disease C, untreated, has a survival rate of 95%.
5. Ask students to reconsider their initial opinions and decisions in light of the new data and decide which disease to treat with which drug.

Assessment Suggestions: Discuss with the class the role of control groups in experimental design and explore how they allow scientists to draw reliable inferences.

Lecture Activity 1.4: Evaluating Scientific Claims in Advertisements

Estimated Time to Complete: 10–30 minutes, depending on how many ads are used

Introduction: We are saturated with stories from the newspaper, magazines, television, and the Internet about the effectiveness of many products, including cold remedies, health supplements, weight-loss products, and so on. These ads are difficult to assess, because we cannot completely determine the quality of the information or the extent to which the financial backers of a product are driving the information being presented. The purpose of this activity is to demonstrate a variety of media resources to the students and ask them their opinions about the effectiveness of the messages portrayed.

Material

- Any source of advertisements will do: magazines, websites, taped TV or radio commercials. Ads can be displayed or reproduced in any convenient fashion. Include ads for various types of products.

Procedures

1. Project or otherwise show students a variety of ads that make claims or promote products.
2. Ask the students to comment on the scientific validity and persuasiveness of the ads. This can be done as a single large discussion or in a small-group format. Discussion questions could include these:
 - Does the ad seem plausible? Why or why not?
 - Are any data cited? Do those data seem complete? Are controls or comparisons included?
 - Are testimonials used? If so, are they convincing? Why or why not?
 - What other information do you wish the ad contained to better help you decide whether to buy the product?
 - Would you buy the product based on this ad? Why or why not?

Assessment Suggestions: A suggested assignment would be for the student to find his or her own advertisement, similar to what the instructor showed in class, and analyze the material for reliability of the information.

In class discussion, poll students to see if any of them have tried or know someone who has used one of the products displayed. Have students compare opinion polling with hypothesis testing based on the use of these methods in advertising and their relative worth as sources of data.

For a group project, students could be assigned a specific health issue (e.g., cold remedies, arthritis relief) and asked to find conflicting testimonials or anecdotes from media sources. Each group could present to the rest of the class their findings on that particular issue and discuss what a primary resource indicates about the subject as compared to secondary sources.

Are We Alone in the Universe?

Water, Biochemistry, and Cells

OBJECTIVES

Teaching Goals

This chapter introduces an invisible world that students often find difficult to grasp. Descriptive imagery and graphic representations are the basis for understanding atoms and cells. As a teacher, you must connect this invisible world to the visible world that students can see. Carbohydrates, proteins, and fats make more sense when seen on a food label, for example. Cells become real when connected to forensic evidence left at the scene.

Student Goals

By the end of this chapter, students should be able to accomplish the following learning objectives:

- Describe the properties associated with living organisms.
- List the components of water and some of the properties that make it important in living organisms.
- Describe how atomic structure affects chemical bonding.
- Compare and contrast hydrogen, covalent, and ionic bonds.
- Discuss the importance of carbon in living organisms.
- Describe the structure of carbohydrates, proteins, lipids, and nucleic acids and the roles these macromolecules play in cells.
- Compare and contrast prokaryotic and eukaryotic cells.
- Describe the structure and functions of cell membranes.
- Describe the structure and function of the subcellular organelles.
- Provide a general summary of the theory of evolution.

LECTURE OUTLINE

2.1 What Does Life Require?

- A. All organisms operate using the same biological molecules, are composed of cells, and use feedback mechanisms to maintain homeostasis. **(Figure 2.1)**
- B. Atoms and subatomic particles play a role in chemical reactions and bond formation. **(Figure 2.3)**
- C. Water is essential to life, and its properties are based on its polarity. **(Figure 2.4, Figure 2.5, Figure 2.6)**

Lecture Activity 2.1

Lecture Activity 2.2

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- D. Acids, bases, and salts play a role in chemical reactions.
- E. Life on Earth is based on the carbon atom and its bonding capabilities to form macromolecules (carbohydrates, proteins, lipids, and nucleic acids). (**Figure 2.9, Figure 2.10, Figure 2.11, Figure 2.12, Figure 2.13, Figure 2.14, Figure 2.15**)

2.2 Life on Earth

- A. Cells are the basic unit of life on Earth.
- B. Cells are either prokaryotic or eukaryotic. Prokaryotic cells, such as bacteria, do not have a separate nucleus and are less complex than eukaryotic cells. (**Figure 2.17**)
- C. All cells have a variety of parts that work together to enable them to function in particular environments. (**Table 2.1**)
Lecture Activity 2.3
Lecture Activity 2.4
- D. The diversity of cells and organisms is based on evolutionary theory. (**Figure 2.19**)

Lecture Activity 2.1: Understanding Chemical Bonding

Estimated Time to Complete: 10–20 minutes

Introduction: This activity is a basic introduction to the concept of chemical bonding. It is a short, in-class exercise that can follow the first discussion of atomic structure and bonding. Students will take the number of their birth month as their atomic number. Given this information, they will be able to determine the configuration of their electrons. They will then be able to determine how they, as atoms, will interact with other atoms, if at all. Students will form “bonds” with other students to make ions or molecules.

Material

- Lecture Activity 2.1 Handout: Understanding Chemical Bonding

Procedures

1. Briefly explain to the students what they will be expected to do on the worksheet.
2. Each student will complete a worksheet alone, but you can allow students to work together if you like.
3. Pass out worksheets, and circulate around the room to ensure that students get started correctly; answer questions that come up.
4. After they know which type of partner they need, encourage students to get out of their seats and walk around to find someone suitable.

Assessment Suggestions: Collect and grade the handouts. Use this assignment as the basis of a homework or test question.

Name: _____

Date: _____

Instructor: _____

Course Section: _____

Lecture Activity 2.1 Handout: Understanding Chemical Bonding

1. What is the number of your birth month? (January = 1, February = 2, etc.) This is your atomic number.
2. Draw your appearance as an atom. Around the nucleus, put the correct number of electrons in their correct valence shells.
3. Which type of bond would you like to form to become stable?
4. Find one or more other students to bond with so that you form a stable bond. Draw the ion or molecule that you all formed.
5. Explain why you are stable in this configuration.

Lecture Activity 2.2: Molecular Twenty Questions

Estimated Time to Complete: 10–20 minutes

Introduction: In this activity, students get an opportunity to test their knowledge of biological molecules. The students will form groups of three to four. Each student will secretly receive the name of a biological molecule. The other students in the group will then take turns asking yes or no questions until one of them is able to correctly guess the identity of the molecule in question.

Material

- Names of different molecules on pieces of paper cut from the bottom of this page.

Procedures

- Divide students into groups of three to four. Explain how the game will work, and have the students determine who will go first in each group.
- Pass around a hat or box containing small folded pieces of paper each with the name of a molecule. In a small class, you can pass out the papers initially to only one student in each group, and then have other students come up for another piece after each round. In a large class, have everyone take a paper at the same time.
- The person going first in each group silently reads his or her molecule's name, and then the other group members take turns asking a yes or no question to try to identify the molecule.
- When students have a guess about the molecule, they must wait until it is their turn to say it.
- Other members of the group then answer questions about their molecules until all members have had a turn, or until the allotted time is up.

Assessment Suggestion: You may have the students keep track of how many molecules they guess correctly, and give out a grade or extra credit for those who score above a certain number.

Molecule Names

DNA	Glucose	Cholesterol
Water	Phospholipid	Polysaccharide
Fat	Sucrose	Protein
Amino acid	Nucleotide	Water

Lecture Activity 2.3: The Cell as a City

Estimated Time to Complete: 5–10 minutes

Introduction: Cells, because they are capable of independent life, must be able to carry out all of the processes necessary for life within themselves. In some ways they resemble cities; the functions of organelles can be compared to utilities and institutions that students are familiar with in the macroscopic world.

Material

- None.

Procedures

After discussing the functions of various organelles, ask students which organelle is most similar to the following institutions:

- A public library (the **nucleus** stores information)
- The U.S. Postal Service (the **Golgi apparatus** sorts and delivers materials)
- A water tower (**vacuoles** store water)
- A coal-burning power plant (**mitochondria** convert fuel to energy)
- A solar power plant (**chloroplasts** convert sunlight to energy)
- Manufacturing plants (**ribosomes** and **endoplasmic reticulum** build proteins)
- A moving company (**centrioles** facilitate movement of chromosomes during mitosis)

Assessment Suggestions: Use similar comparisons on test questions.

Lecture Activity 2.4: Cells Alive!

Estimated Time to Complete: 5–10 minutes

Introduction: The concepts of living cells are central to every biology course. However, it is difficult for students to discuss life and its dynamics when they are exposed to only two-dimensional, static pictures. This lecture activity brings the concepts of cells to life by letting students watch cancer and bacterial cells divide in real time. This activity can be used to discuss differences between prokaryotic and eukaryotic cells, cell structure, and the characteristics of life.

Material

- A computer with projector capabilities and an Internet connection.

Procedures

1. In the classroom, project the following website: <http://www.cellsalive.com>.
2. Many areas related to cells can be explored at this website, but a favorite is “Cell Cam.” You can find this link along the left side of the “CELLS alive!” home page.
3. You will see two pictures: one of human cancer cells dividing and one of bacterial cells dividing. When you click on one of these pictures, you will link to a page that has three picture frames. The first picture was taken earlier in the day; the time this picture was taken is shown at the top of the picture. The picture on the far right is the most recent picture taken; its time is also shown. The middle picture will look like the left picture until you change it by clicking the 1 and 2 buttons below the pictures.
4. The “CELLS alive!” website maintains a live link to a camera that takes pictures of these cells every few minutes. You can click on the +1 and +5 buttons below the pictures to view the pictures that were

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taken between the first and the current pictures. In this way, you can watch the cells divide as you click through the sequence of pictures taken that day.

5. If you remain on this site, you can watch the next pictures taken. Students really love the idea of the pictures being taken “live” right in front of them. They are actually witness to cell division events.
6. Demonstrate the division of human cells and bacterial cells by clicking consecutively through an entire day of pictures. Have students note how fast each cell type divides, what the size and shape of the cells are, and how the cells react to one another in the Petri plate.
7. Have students draw and label some of what they see. Are there internal structures evident in the pictures of these cells? Why or why not? How are these pictures different from the figures in the textbook? Can students visualize these cells dividing in this way in the human body?

Assessment Suggestions: Students may turn in their drawings and observations. There are also many ideas for follow-up activities on this website.

This could also be done as an online assignment. Provide students with the link and then create an assignment using a variety of questions.

Is It Possible to Supplement Your Way to Better Health?

Nutrients and Membrane Transport

OBJECTIVES

Teaching Goals

Many students read food labels but they don't understand the significance of the information provided on those labels. In this chapter you should clearly differentiate between carbohydrates, fats, and proteins. Your goal is to help students recognize and make healthful food choices by understanding food labels. The concept of membrane transport is often difficult for students to grasp. Use visual aids and animations to enhance their understanding of osmosis, diffusion, and active transport.

Student Goals

By the end of this chapter students should be able to accomplish the following learning objectives:

- Describe the role of nutrients in the body.
- Describe the function of the water in the body.
- Describe the major dietary macronutrients and discuss the functions of each.
- List the major dietary micronutrients and describe their functions.
- Describe the structure and function of the plasma membrane.
- Distinguish between passive transport and active transport.
- Describe the processes of endocytosis and exocytosis.

LECTURE OUTLINE

3.1 Nutrients

- A. Macronutrients such as water, carbohydrates, proteins, and fats are nutrients that are required in large amounts.
- B. Water is required to carry nutrients throughout the body and maintain an even temperature.
- C. Carbohydrates can be either simple or complex; both types serve different functions in the body. **(Figure 3.1)**
- D. Proteins are made of amino acids and are found in a variety of foods. **(Figure 3.2)**
- E. Fats serve as a source of energy as well as protection for organs. They are classified as saturated, unsaturated, or trans fats. **(Figure 3.4)**

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- F. Micronutrients, which include vitamins, minerals, and antioxidants, are essential parts of a healthy diet. (**Table 3.1, Table 3.2, Table 3.3**)
 - G. Whole foods contain more nutrients, vitamins, and minerals than do processed foods.
- Lecture Activity 3.1**

3.2 Transport Across Membranes

Lecture Activity 3.2

- A. Passive transport does not require an input of energy and can occur by diffusion, facilitated diffusion, or osmosis. (**Figure 3.5, Figure 3.6**)

Lecture Activity 3.3

- B. Active transport requires energy and uses proteins to move molecules from areas of lower concentration to areas of higher concentration. (**Figure 3.7**)
- C. Large molecules are taken into the cell by endocytosis and leave the cell by exocytosis. (**Figure 3.8**)

Lecture Activity 3.1: Reading Nutrition Labels

Estimated Time to Complete: 15 minutes

Introduction: Students are repeatedly told that they need to eat a balanced diet, yet many are not able to evaluate the foods they eat. It is important that students learn how to read nutrition labels on the foods they eat and use this information to evaluate a product for its nutritional content.

Materials

- Three nutrition labels: two from “healthy” foods and one from a “junk” food. Copy one of the “healthy” labels onto a PowerPoint slide or a transparency. Copy the other two labels side by side onto a PowerPoint slide or a transparency.
- Lecture Activity 3.1 Handout: Reading Nutrition Labels

Procedures

1. Take 5 minutes to describe the parts of a nutrition label using the “healthy” food label on the transparency or slide. Include the following important points:
 - A. %DV: If a nutrition label displays a 6% DV for Total Carbohydrates, this means that by eating a single serving of this food, you will be consuming 6% of your daily allotment for carbohydrates. This figure is based on a 2000-calorie diet.
 - B. Total Fat value minus Saturated Fat value equals Unsaturated Fat value.
 - C. Total Carbohydrate value minus Fiber value minus Sugar value equals Complex Carbohydrate value.
 - D. Serving size is important. Most Americans eat more than a serving-size portion at one sitting.
 - E. Many labels also include the following information:
 - 1 g Fat = 9 calories.
 - 1 g Carbohydrates = 4 calories.
 - 1 g Protein = 4 calories.
2. Show the second overhead transparency or slide, and direct the students to answer the questions on the handout in groups of two or three. Circulate among the groups to ensure that each group is on task. Answer questions to keep students moving forward.
3. Have a few students give oral reports on their findings and conclusions.

Assessment Suggestions: Have students write their answers and calculations on the handout provided, and collect their group's work. Evaluations may not be easily given for points, but this will be valuable material with which the instructor can assess the level of student understanding of this topic.

Ask students to choose a food label from a food they consider "unhealthy" and one from a "healthy" food. These labels should come from foods they consume in their personal diet. Have students use the food label information to support their position in a one-page paper.

You can do this assignment online by scanning in different labels and placing the questions online for them to answer. You could also do a group project and have students find five labels and then explain them in a one-page group paper.

Name: _____

Date: _____

Instructor: _____

Course Section: _____

Lecture Activity 3.1 Handout: Reading Nutrition Labels

Answer the following questions in your small groups. Be prepared to share your answers in a class discussion.

1. Circle the main type of carbohydrate present in the foods shown.
fiber simple sugar complex carbohydrate
2. Why does the kind of carbohydrate matter when eating a healthy diet?
3. Are either of these foods high in saturated fat? Yes or No
4. Do you see any “hydrogenated” fats (a source of trans-fatty acids) listed in the ingredients? Yes or No
5. Do either of these foods have a good balance of fats, proteins, and carbohydrates? Yes or No
6. If your answer to question 5 was no, then which kinds of foods would you need to eat to balance out the rest of your day’s diet?
7. Is one of these foods healthier than the other? Yes or No
8. If the answer to question 7 is yes, write the name of the food.
9. What criteria did you use to determine the food’s “healthiness”?

Lecture Activity 3.2: Interaction of Hydrophilic and Hydrophobic Substances

Estimated Time to Complete: 15 minutes

Introduction: To understand the structure and function of biological membranes, students need to understand some basic characteristics of biological molecules. This demonstration will reinforce the differences between hydrophilic and hydrophobic substances, and show that they create two distinct layers when mixed. When a hydrophilic solute is added, it will be more soluble in the water layer. In contrast, a hydrophobic solute will be more soluble in the lipid layer. This information could then lead to a discussion regarding the solubility or permeability of solutes in these layers.

Materials

- 30 ml of water
- 100-ml graduated cylinder or similar tall, thin container
- Methylene blue, or blue or green food-coloring dye
- 30 ml of vegetable oil
- Lipid-soluble dye such as Sudan red (in powder form)
- Parafilm or plastic wrap
- Lecture Activity 3.3 Handout: Interaction of Hydrophilic and Hydrophobic Substances

Procedures

1. Place 30 ml of water in the graduated cylinder.
2. Add methylene blue or food coloring; swirl.
3. Carefully pour 30 ml of vegetable oil on top so that two layers exist.
4. Add Sudan red dye.
5. Cover top of graduated cylinder with parafilm or plastic wrap. Shake the container, and place it on a desk to settle during the class period.
6. At the end of the period, discuss what is observed, using the assessment questions on the handout.

Assessment Suggestion: Use the handout provided for class discussion or individual assessment. You could also have students add some detergent to the graduated cylinder after step 5 and shake it to demonstrate how detergent allows hydrophobic and hydrophilic molecules to temporarily mix.

Name: _____

Date: _____

Instructor: _____

Course Section: _____

Lecture Activity 3.2 Handout: Interaction of Hydrophilic and Hydrophobic Substances

Answer the following questions in your small groups. Be prepared to share your answers in a class discussion.

1. Describe what occurs when the oil and water are mixed together.
2. After the solution has been mixed and allowed to settle, describe what you observe.
3. Explain why this result occurred.
4. Relate this demonstration to what you know about membrane structure.
5. Explain why hydrophilic compounds such as amino acids and sugars require facilitated diffusion or active transport to cross a cell membrane.