Chapter 2
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy

1. What is the molecular formula for the following structure?
$\stackrel{\circ}{\circ}$
A. $\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}$
B. $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}$
C. $\mathrm{C}_{4} \mathrm{H}_{10} \mathrm{O}$
D. $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{O}$

Ans: C
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy
2. Which of the following structures is of a compound with a molecular formula of $\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}$ ?

1

2

3
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OCH}_{3}$
4
A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy
3. Which of the following is the correct condensed structure for the following structure?

$\mathrm{CH}_{3} \mathrm{CHCH}_{2} \mathrm{OH}$
1
$\left(\mathrm{CH}_{3}\right)_{2} \mathrm{CHCH}_{2} \mathrm{OH}$
2
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$
3
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$
4
A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy
4. Which of the following is the correct Lewis structure for the following structure?


1

2

3

A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy
5. Which of the following is the correct bond-line structure for $\left(\mathrm{CH}_{3}\right)_{4} \mathrm{C}$ ?


3

4
4
1


2
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Medium
6. Draw the bond-line structure for $\left(\mathrm{CH}_{3}\right)_{4} \mathrm{C}$ ?

Ans:


## Topic: Molecular Representations

Section: 2.1
Difficulty Level: Easy
7. Which of the following is the correct molecular formula for $\left(\mathrm{CH}_{3}\right)_{4} \mathrm{C}$ ?
$\mathrm{C}_{2} \mathrm{H}_{12}$
1
$\mathrm{C}_{5} \mathrm{H}_{12}$
2
$\mathrm{C}_{4} \mathrm{H}_{12}$
3
$\mathrm{C}_{3} \mathrm{H}_{9}$
4
A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy
8. Which of the following is the correct Lewis structure for $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NH}_{2}$ ?



3

4
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Molecular Representations
Section: 2.1
Difficulty Level: Easy
9. Which of the following is a valid Lewis structure for a compound with the molecular formula of $\mathrm{C}_{4} \mathrm{H}_{11} \mathrm{~N}$ ?

1

2

3

A. 1
B. 2
C. 3
D. 4

Ans: A

Section: 2.1
Difficulty Level: Medium
10. Draw a valid Lewis structure for the following compound.


Ans:


Topic: Molecular Representations
Section: 2.1
Difficulty Level: Hard
11. Draw a valid Lewis structure for the following compound.
$\mathrm{H}_{3} \mathrm{CCOH}$

Ans:


Topic: Molecular Representations
Section: 2.1
Difficulty Level: Hard
12. Draw a valid Lewis structure for the following compound. $\mathrm{H}_{3} \mathrm{CSOCH}_{3}$

Ans:


Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
13. Which of the following bond-line structures are of the same compound?


1


2


3


4
A. 1 and 2
B. 2 and 3
C. 3 and 4
D. 1 and 3

Ans: D
Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
14. Which of the following bond-line structures are of the same compound?

1

2

3

4
A. 1 and 2
B. 2 and 3
C. 3 and 4
D. 2 and 4

Ans: D
Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
15. In the following reaction, how many H's do you add or lose?

ค $\qquad$入
A. Add 1
B. Add 2
C. Lose 1
D. Lose 2
E. No change

Ans: D
Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
16. In the following reaction, how many H's do you add or lose?

A. Add 1
B. Add 2
C. Lose 1
D. Lose 2
E. No change

Ans: B

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Medium
17. In the following reaction, how many H's do you add or lose?

$\qquad$

A. Add 1
B. Add 2
C. Lose 1
D. No change

Ans: D

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Medium
18. In the following reaction, how many H's do you add or lose?

A. Add 1
B. Add 2
C. Lose 1
D. No change

Ans: D

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
19. In the following reaction, how many H's do you add or lose?

A. Add 1
B. Add 2
C. Lose 1
D. No change

Ans: D

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
20. In the following reaction, how many H's do you add or lose?

A. Add 1
B. Add 2
C. Lose 1
D. Lose 2

Ans: D

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
21. How many H's are on the indicated carbon?

A. 1
B. 2
C. 3
D. 4

Ans: C

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
22. How many H's are on the indicated carbon?

A. 1
B. 2
C. 3
D. 0

Ans: D
Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Easy
23. How many H's are on the indicated carbon?

A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Medium
24. How many H's are on the indicated carbon?

A. 1
B. 2
C. 3
D. 4

Ans: A
Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Medium
25. How many H's are on the indicated carbon?

A. 1
B. 2
C. 3
D. 0

Ans: D

Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Medium
26. Convert $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2}\right)_{2}$ into a bond-line structure.

Ans:
Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Hard
27. Draw bond-line structure for all of the isomers with a molecular formula of $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{O}$.

Ans:


Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Hard
28. Draw bond-line structure for all of the isomers with a molecular formula of $\mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}$.

Ans:



Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Hard
29. Draw a line-bond structure for a compound with a molecular formula of $\mathrm{C}_{4} \mathrm{H}_{12}$ and one carbon with no hydrogens attached to it.

Ans:


Topic: Bond-Line Structures
Section: 2.2
Difficulty Level: Hard
30. Draw a line-bond structure for a compound with a molecular formula of $\mathrm{C}_{5} \mathrm{H}_{14}$ and one carbon with only one hydrogen attached to it.

Ans:


Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
31. Which of the following structures contains an alcohol?


1


2


3


4
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
32. Which of the following structures contains an alkene?

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
33. Which of the following structures contains an amine?

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: D
Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
34. Which of the following structures contains a ketone?

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
35. Which of the following structures contains an aromatic ring?


1


2


3


4
A. 1
B. 2
C. 3
D. 4

Ans: D

Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
36. Which of the following structures contains an alkyne?

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
37. Which of the following structures contains a nitrile?

2

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: A
Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Easy
38. Which of the following structures contains an amide?

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Hard
39. Draw all the isomers with a molecular formula of $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}$ and label all the functional groups.


Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Hard
40. Draw the isomers with a molecular formula of $\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}_{2}$ and label the functional groups.
Ans:


Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Hard
41. Identify the functional groups in the following compound.


Ans:


Topic: Identifying Functional Groups
Section: 2.3
Difficulty Level: Hard
42. Identify the functional groups in the following compound.


Ans:


Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
43. Calculate the formal charge on the indicated atom.

A. +1
B. +2
C. -1
D. -2

Ans: A
Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
44. Calculate the formal charge on the indicated atom.

A. +1
B. +2
C. -1
D. -2

Ans: A
Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
45. Calculate the formal charge on the indicated atom.

A. +1
B. +2
C. -1
D. -2

Ans: A
Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
46. Calculate the formal charge on the indicated atom.

A. +1
B. +2
C. -1
D. 0

Ans: D

Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
47. Calculate the formal charge on the indicated atom.

A. +1
B. +2
C. -1
D. 0

Ans: D
Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
48. Calculate the formal charge on the indicated atom.

A. +1
B. +2
C. -1
D. -2

Ans: A
Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
49. Determine the number of hydrogens on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: A
Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
50. Determine the number of hydrogens on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: A

Topic: Formal Charges
Section: 2.4
Difficulty Level: Easy
51. Determine the number of hydrogens on the indicated atom.

A. 1
B. 2
C. 3
D. 0

Ans: D
Topic: Formal Charges
Section: 2.4
Difficulty Level: Medium
52. Determine the formal charges on every atom in the following structure.


Ans:


Topic: Formal Charges
Section: 2.4
Difficulty Level: Hard
53. Draw a valid Lewis structure, with all lone pairs and formal charges for the azide anion $\left(\mathrm{N}_{3}{ }^{-}\right)$.
Ans:


Topic: Identifying Lone Pairs
Section: 2.5

Difficulty Level: Easy
54. Determine the number of lone pairs of electrons on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Identifying Lone Pairs
Section: 2.5
Difficulty Level: Easy
55. Determine the number of lone pairs of electrons on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Identifying Lone Pairs
Section: 2.5
Difficulty Level: Easy
56. Determine the number of lone pairs of electrons on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: A

Topic: Identifying Lone Pairs
Section: 2.5
Difficulty Level: Easy
57. Determine the number of lone pairs of electrons on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Identifying Lone Pairs
Section: 2.5
Difficulty Level: Easy
58. Determine the number of lone pairs of electrons on the indicated atom. $\rightarrow$
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Identifying Lone Pairs
Section: 2.5
Difficulty Level: Easy
59. Determine the number of lone pairs of electrons on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Identifying Lone Pairs
Section: 2.5
Difficulty Level: Easy
60. Determine the number of lone pairs of electrons on the indicated atom. $\xrightarrow[\sim]{\sim}$
A. 1
B. 2
C. 3
D. 4

Ans: A
Topic: Identifying Lone Pairs
Section: 2.5

Difficulty Level: Easy
61. Determine the number of lone pairs of electrons on the indicated atom.

A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Lone Pairs
Section: 2.5
Difficulty Level: Medium
62. Draw all of the lone pairs on the following compound.


Ans:


Topic: Lone Pairs
Section: 2.5
Difficulty Level: Medium
63. Draw all of the lone pairs on the following compound.


Ans:


Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Easy
64. For the following compound, is the indicated bond up or down?

A. Up
B. Down

Ans: A

Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Easy
65. For the following compound, is the indicated bond up or down?

A. Up
B. Down
C. Neither

Ans: C

Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Easy
66. For the following compound, is the indicated bond up or down?

A. Up
B. Down
C. Neither

Ans: B

Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Easy
67. Which of the following structures is the same as structure A?



1
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Easy
68. Which of the following structures is a Fischer projection?
$\sim$
1

2
$\square$
3
$\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}$
4
A. 1
B. 2
C. 3
D. 4

Ans: B
Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Easy
69. Which of the following structures is a Haworth projection?

1

2

3
$\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}$
4
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Medium
70. Draw an example of a Fischer projection.

Ans: many!
Topic: Three-Dimensional Bond-line Structures
Section: 2.6
Difficulty Level: Medium
71. Circle the atoms which are pointing up in the following compound.


Ans:


Topic: Introduction to Resonance
Section: 2.7
Difficulty Level: Easy
72. Which of the following pairs of structures are resonance structures?
— N



1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Introduction to Resonance
Section: 2.7
Difficulty Level: Easy
73. Which of the following pairs of structures are resonance structures?


1
= N



2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: C

Topic: Introduction to Resonance
Section: 2.7
Difficulty Level: Hard
74. When do we use resonance structures?

Ans: When the structure of a compound cannot be adequately described by one single structure.

Topic: Introduction to Resonance
Section: 2.7
Difficulty Level: Hard
75. Draw all valid resonance structures for the following anion.


Ans:


Topic: Curved Arrows
Section: 2.8
Difficulty Level: Medium
76. Draw in the appropriate arrows for the following resonance pairs.


Ans:


Topic: Curved Arrows
Section: 2.8
Difficulty Level: Easy
77. In the following structure, the arrow indicates that electrons are moving to what atom?

A. C
B. O
C. H
D. N

Ans: B

Topic: Curved Arrows
Section: 2.8
Difficulty Level: Easy
78. In the following structure, the arrow indicates that electrons are moving to what atom?

A. C
B. O
C. H
D. N

Ans: B
Topic: Curved Arrows
Section: 2.8
Difficulty Level: Easy
79. In the following structure, the arrow indicates that electrons are moving to what atom?

A. C
B. O
C. H
D. N

Ans: A
Topic: Curved Arrows
Section: 2.8
Difficulty Level: Hard
80. Draw in the appropriate arrows for the following resonance pairs.


Ans:


Topic: Curved Arrows
Section: 2.8
Difficulty Level: Hard
81. Draw in the appropriate arrows for the following resonance pairs.


Ans:


Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Easy
82. Which of the following is a correct resonance structure of A ?


A


1


2


3


4
A. 1
B. 2
C. 3
D. 4

Ans: D

Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Easy
83. Which of the following is a correct resonance structure of A?

A

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Easy
84. Which of the following is a correct resonance structure of A?

A

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Easy
85. Which of the following is a correct resonance structure of A ?
$\rangle=\mathrm{N}$
A

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: D
Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Medium
86. Draw the resonance structure that is indicated by the following arrows.
$\overbrace{\mathrm{O}_{\mathrm{N}^{\prime}} \mathrm{O}^{+}} \mathrm{O}^{\ominus}$

Ans:


Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Medium
87. Draw the resonance structure that is indicated by the following arrows.


Ans:


Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Medium
88. Draw the resonance structure that is indicated by the following arrows.


Ans:


Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Medium
89. Draw the resonance structure that is indicated by the following arrows.


Ans:


Topic: Formal Charges in Resonance Structures
Section: 2.9
Difficulty Level: Medium
90. Draw the resonance structure that is indicated by the following arrows.


Ans:


Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Easy
91. Which of the following is a correct resonance structure of A ?

A

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: A
Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Easy
92. Which of the following is a correct resonance structure of A?

A

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: B

Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Medium
93. Draw a resonance structure of A .


A
Ans:


Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Easy
94. Which of the following is a correct resonance structure of A?

A

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Easy
95. Which of the following is a correct resonance structure of A ?


A


1


2


3


4
A. 1
B. 2
C. 3
D. 4

Ans: C

Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Hard
96. Draw a resonance structure of A.


A
Ans:


Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Easy
97. Which of the following is a correct resonance structure of A ?

A

1

2

3

4
A. 1
B. 2
C. 3
D. 4

Ans: C
Topic: Pattern Recognition
Section: 2.10
Difficulty Level: Medium
98. Draw a resonance structure of A.


A
Ans:


Topic: Assessing Importance
Section: 2.11
Difficulty Level: Easy
99. Is the following a valid resonance pair?

A. Yes
B. No

Ans: B

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Easy
100. Is the following a valid resonance pair?

A. Yes
B. No

Ans: A

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Easy
101. Is the following a valid resonance pair?

A. Yes
B. No

Ans: B

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Easy
102. Which of the following resonance structures is the most significant?


1


2


3
A. 1
B. 2
C. 3

Ans: C

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Easy
103. Which of the following resonance structures is the most significant?




1
2

3
A. 1
B. 2
C. 3

Ans: C

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Hard
104. Draw the resonance structures of the nitrite anion $\left(\mathrm{NO}_{2}{ }^{-}\right)$and indicate the most important one.
Ans: $\mathrm{O}=\mathrm{N}-\mathrm{O}^{\ominus}$

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Hard
105. Draw the resonance structures of the cyanide anion $\left(\mathrm{CN}^{-}\right)$, indicating the most important one.
Ans: ${ }^{\ominus} \mathrm{C} \equiv \mathrm{N}$ :

Topic: Assessing Importance
Section: 2.11
Difficulty Level: Medium
106. Draw the resonance structures of the following anion.


Ans:


Topic: Assessing Importance
Section: 2.11
Difficulty Level: Hard
107. Draw the resonance structures of the following cation and indicate the most important one.


Ans:


Topic: Assessing Importance
Section: 2.11
Difficulty Level: Easy
108. Which of the following resonance structures of dimethylaniline is the most significant?



2


3
A. 1
B. 2
C. 3

Ans: C

Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Easy
109. The lone pair in the following compound is:

A. localized
B. delocalized

Ans: A
Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Easy
110. The lone pairs in the following compound are:
$\widehat{\mathrm{O}}$
A. localized
B. delocalized

Ans: B
Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Easy
111. The indicated lone pair in the following compound is:

A. localized
B. delocalized

Ans: A

Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Easy
112. The indicated lone pair in the following compound is:

A. localized
B. delocalized

Ans: B

Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Easy
113. The indicated lone pair in the following compound is:

A. localized
B. delocalized

Ans: A

Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Hard
114. Draw in all localized lone pairs on the following structure.


Ans:


Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Hard
115. Draw in all localized lone pairs on the following structure.


Ans:


Topic: Delocalized and Localized Lone Pairs
Section: 2.12

## Difficulty Level: Hard

116. Draw in all localized lone pairs on the following structure.


Ans:


Topic: Delocalized and Localized Lone Pairs
Section: 2.12
Difficulty Level: Hard
117. Draw in all delocalized lone pairs on the following structure.


Ans:


