Copyright Cengage Learning. Powered by Cognero.

1. Approximately how long is a C-C single bond of an alkane? a. 111 pm b. 134 pm c. 142 pm d. 153 pm ANSWER: d **POINTS:** 1 QUESTION TYPE: Multiple Choice HAS VARIABLES: False DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM 2. What is the approximate C-C-C bond angle in propane? a. 90° b. 109° c. 120° d. 180° ANSWER: b **POINTS:** 1 QUESTION TYPE: Multiple Choice HAS VARIABLES: False DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM 3. What is the name of the linear hydrocarbon with the molecular formula C₇H₁₆? a. hexane b. heptane c. decane d. undecane ANSWER: b **POINTS:** 1 QUESTION TYPE: Multiple Choice HAS VARIABLES: False DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM 4. What is the name of the linear hydrocarbon with the molecular formula $C_{11}H_{24}$? a. heptane b. decane c. undecane d. eicosane ANSWER: c **POINTS:** QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 5. How many hydrogen atoms are there in nonane, the linear hydrocarbon with nine carbon atoms?
 - a. 16
 - b. 18
 - c. 20
 - d. 22

ANSWER: c

POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 6. How many hydrogen atoms are there in dodecane, the linear hydrocarbon with twelve carbon atoms?
 - a. 12
 - b. 20
 - c. 24
 - d. 26

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 7. How many constitutional isomers are there with the molecular formula C₄H₁₀?
 - a. 2
 - b. 3
 - c. 4
 - d. 5

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 8. How many constitutional isomers are there with the molecular formula C₅H₁₂?
 - a. 2
 - b. 3
 - c. 4

d. 5

ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 9. How many constitutional isomers are there with the molecular formula C₆H₁₄?
 - a. 3
 - b. 4
 - c. 5
 - d. 8

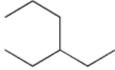
ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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10. What is the IUPAC name of the following compound?



- a. 3-propylpentane
- b. 1,1-diethylpropane
- c. 3-ethylhexane
- d. isooctane

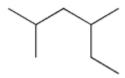
ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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11. What is the IUPAC name of the following compound?



- a. 2-ethyl-4-methylpentane
- b. 2,4-dimethylhexane

c. 3,5-dimethylhexane

d. 1,1,3-trimethylpentane

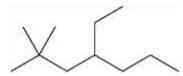
ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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12. What is the IUPAC name of the following compound?



a. 2,2-dimethyl-4-ethylheptane

b. 4-ethyl-2,2-dimethyl-heptane

c. 6,6-dimethyl-4-ethylheptane

d. 4-ethyl-6,6-dimethyl-heptane

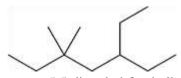
ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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13. What is the IUPAC name of the following compound?



a. 5,5-dimethyl-3-ethylheptane

b. 5-ethyl-3,3-dimethyl-heptane

c. 3,3-dimethyl-5-ethylheptane

d. 3-ethyl-5,5-dimethyl-heptane

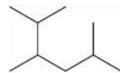
ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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14. What is the IUPAC name of the following compound?



a. 2-isopropyl-5-methylpentane

b. 5-isopropyl-2-methylpentane

c. 2,3,5-trimethylhexane

d. 1,2-diisopropylpropane

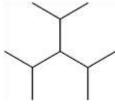
ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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15. What is the IUPAC name of the following compound?



a. 2,4-dimethyl-3-isopropyl-pentane

b. 3-isopropyl-1,5-dimethylpentane

c. 3-isopropyl-2,4-dimethylpentane

d. triisopropylmethane

ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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16. Which of the following compounds has 1°, 2°, 3° and 4° carbon atoms?

- a. hexane
- b. 2-methylhexane
- c. 2,2-dimethylhexane
- d. 2,2,3-trimethylhexane

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 17. Which of the following compounds has only 1° and 3° carbon atoms?
 - a. hexane
 - b. 2-methylpentane
 - c. 3-methylpentane
 - d. 2,3-dimethylbutane

ANSWER: d

POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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18. What is the correct assignment of common names for the following molecules?

a. i = butane; ii = neopentane; iii = isopentane

b. i = neobutane; ii = isobutane; iii = pentane

c. i = butane; ii = isobutane; iii = isopentane

d. i = butane; ii = isobutane; iii = neopentane

ANSWER: d POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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19. What is the correct assignment of common names for the following molecules?

CH₃
$$iii$$
 iii iii a. $i = \text{pentane}$; $ii = \text{isopentane}$; $iii = \text{neopentane}$

b. i = neopentane; ii = isopentane; iii = pentane

c. i = pentane; ii = neopentane; iii = isopentane

d. i = neopentane; ii = pentane; iii = isopentane

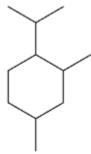
ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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20. What is the IUPAC name of the following compound?



a. 1-isopropyl-4,6-dimethylcyclohexane

b. 1-isopropyl-2,4-dimethylcyclohexane

c. 4-isopropyl-1,3-dimethylcyclohexane

d. 4-isopropyl-1,5-dimethylcyclohexane

ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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21. What is the IUPAC name of the following compound?



a. 1-methylbicyclo[2.2.1]heptane

b. 2-methylbicyclo[2.2.1]heptane

c. 3-methylbicyclo[2.2.1]heptane

d. 4-methylbicyclo[2.2.1]heptane

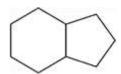
ANSWER: b
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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22. What is the IUPAC name of the following compound?



- a. bicyclo[4.3]nonane
- b. bicyclo[4.3.0]nonane
- c. bicyclo[6.5]nonane
- d. bicyclo[6.5.0]nonane

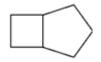
ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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23. What is the IUPAC name for the following compound?



- a. cycloheptane
- b. bicyclo[3.2.0]heptane
- c. bicyclo[5.4]heptane
- d. cyclobutylcyclopentane

ANSWER: b
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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24. What is the IUPAC name for the following compound?



- a. bicyclo[5.4.3]octane
- b. bicyclo[3.2.1]octane
- c. bicyclo[3.2.1]hexane
- d. bicyclo[2.2.1]octane

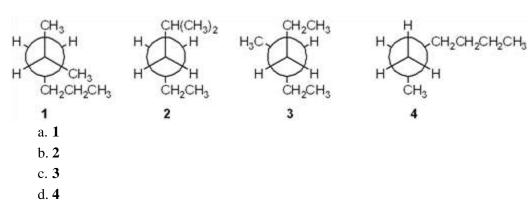
ANSWER: b
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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25. Which of the following Newman projections does *not* represent 2-methylhexane?



ANSWER: c **POINTS:** 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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26. Which of the following Newman projections represents 2,4-dimethylpentane?

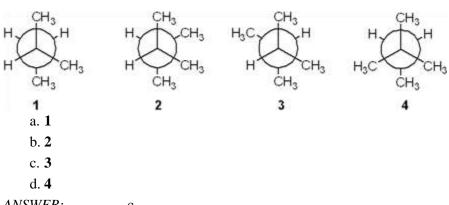
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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27. Which of the following Newman projections represents the most stable conformation of 2,3-dimethylbutane?



ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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28. Which of the following Newman projections represents the most stable conformation of 2-methylbutane?

ANSWER: a POINTS: 1

c. 3 d. 4

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 29. Which of the following cycloalkanes has the most ring strain?
 - a. cyclopropane
 - b. cyclobutane
 - c. cyclopentane
 - d. cyclohexane

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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30. Which of the following cycloalkanes has the least ring strain?

a. cyclopropane

b. cyclopentane

c. cyclohexane

d. cycloheptane

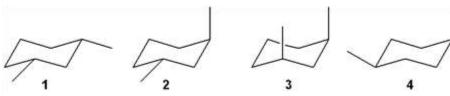
ANSWER: **POINTS:** 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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31. Which of the following structures represents *trans*-1,3-dimethylcyclohexane?



a. 1

b. 2

c. 3

d. 4

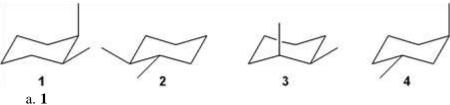
ANSWER: b **POINTS:**

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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32. Which of the following structures represents *trans*-1,2-dimethylcyclohexane?



b. 2

c. 3

d. 4

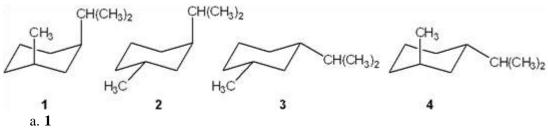
ANSWER: b **POINTS:** 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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33. Which of the following is the most stable conformation of cis-1-isopropyl-3-methylcyclohexane?



b. **2**

c. 3

d. **4**

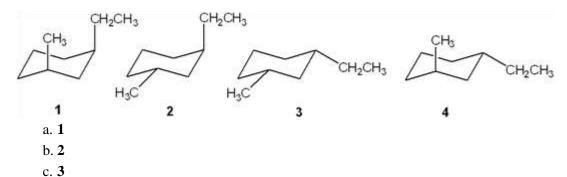
ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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34. Which of the following is the most stable conformation of *trans*-1-ethyl-3-methylcyclohexane?



d. **4**ANSWER:

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 35. Which of the following alkanes has the highest boiling point?
 - a. propane
 - b. butane
 - c. pentane

d. hexane

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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36. Which of the following alkanes has the highest boiling point?

a. 2,3-dimethylbutane

b. 2-methylpentane

c. 3-methylpentane

d. hexane

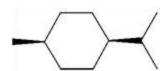
ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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37. What is the IUPAC name of the following compound?



a. trans-1-isopropyl-4-methylcyclohexane

b. cis-1-isopropyl-4-methylcyclohexane

c. cis-2-isopropyl-5-methylcyclohexane

d. cis-1-tert-butyl-4-methylcyclohexane

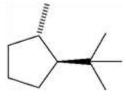
ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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38. What is the IUPAC name of the following compound?



a. trans-1-isopropyl-4-methylcyclopentane

b. cis-1-tert-butyl-2-methylcyclopentane

c. trans-1-tert-butyl-2-methylcyclopentane

d. cis-1-isopropyl-2-methylcyclopentane

ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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39. Which one of the following structures represents a different compound from the other three?

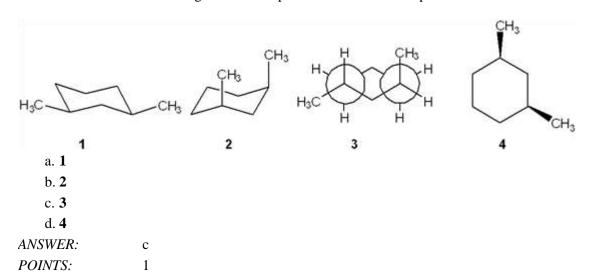
ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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40. Which one of the following structures represents a different compound from the other three?



QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 41. In which of the following compounds are all of the carbon atoms in the same plane?
 - a. cyclopropane
 - b. cyclobutane
 - c. cyclopentane
 - d. cyclohexane

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 42. Which of the following compounds can adopt a chair conformation in which there are no axial methyl groups?
 - a. 1,1-dimethylcyclohexane
 - b. *cis*-1,2-dimethylcyclohexane
 - c. trans-1,2-dimethylcyclohexane
 - d. cis-1,3-dimethylcyclohexane

ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 43. Which of the following compounds can adopt a chair conformation in which there are no axial methyl groups?
 - a. cis-1,2-dimethylcyclohexane
 - b. cis-1,3-dimethylcyclohexane
 - c. trans-1,3-dimethylcyclohexane
 - d. cis-1,4-dimethylcyclohexane

ANSWER: b
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 44. Which of the following statements is not true regarding the conformation of substituted cyclohexanes?
 - a. ring inversion of cyclohexane between two chair conformations takes place via a boat conformation
 - b. substituted cyclohexanes are destabilized by 1,3-diaxial interactions
 - c. the boat conformation of cyclohexane is usually more stable than the chair conformation

d. the relative amount of two conformations of substituted cyclohexanes can be determined from the difference in strain energy ANSWER: c **POINTS:** 1 QUESTION TYPE: Multiple Choice HAS VARIABLES: False DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM 45. What is the approximate dihedral angle between the two chlorine atoms in cis-1,2-dichlorocyclohexane? a. 0° b. 60° c. 120° d. 180° ANSWER: b **POINTS:** 1 QUESTION TYPE: Multiple Choice HAS VARIABLES: False DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM 46. What is the approximate dihedral angle between the two chlorine atoms in the diequatorial conformation of trans-1,2dichlorocyclohexane? a. 0° b. 60° c. 120°

d. 180°

ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 47. What is the approximate dihedral angle between the two chlorine atoms in the diaxial conformation of *trans*-1,2-dichlorocyclohexane?
 - a. 0°
 - b. 60°
 - c. 120°
 - d. 180°

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 48. Which of the following is *not* true regarding the properties of alkanes?
 - a. alkanes are nonpolar
 - b. alkanes burn in air to give H₂O and CO₂
 - c. alkanes are highly miscible with water
 - d. the strongest intermolecular force between alkane molecules is the van der Waals interaction

ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 49. Which of the following undergoes the most exothermic combustion?
 - a. octane
 - b. 2-methylheptane
 - c. 2,2-dimethylhexane
 - d. 2,2,3,3-tetramethylbutane

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 50. How many moles of molecular oxygen (O_2) are consumed in the complete combustion of one mole of octane (C_8H_{18}) ?
 - a. 12.5
 - b. 13
 - c. 17
 - d. 26

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 51. How many moles of molecular oxygen (O_2) are consumed in the complete combustion of one mole of hexane (C_6H_{14})?
 - a. 6
 - b. 9.5

c. 12.5

d. 14

ANSWER: b
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 52. Which of the following statements is *not* true?
 - a. Combustion of an alkane is an exothermic reaction.
 - b. The heat of combustion of propane is three times that of methane.
 - ^{c.} The constitutional isomers of C₇H₁₆ have different heats of combustion from one another
 - d. The products of combustion of an alkane are H₂O and CO₂.

ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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53. Which of the following is the steroid nucleus?

POINTS:

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 54. Which of the following cycloalkanes has the largest heat of combustion?
 - a. cyclopropane
 - b. cyclobutane
 - c. cyclopentane
 - d. cyclohexane

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 55. Which of the following cycloalkanes has the largest heat of combustion per carbon atom?
 - a. cyclopropane
 - b. cyclopentane
 - c. cyclohexane
 - d. cycloheptane

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 56. Which of the following cycloalkanes has the smallest heat of combustion per carbon atom?
 - a. cyclopropane
 - b. cyclopentane
 - c. cyclohexane
 - d. cycloheptane

ANSWER: c

POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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57. Which of the following structures is different from the other three?

a. 1

b. **2**

c. 3

d. **4**

ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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58. Which of the following structures is different from the other three?

a. 1

b. **2**

c. 3

d. 4

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

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- 59. Which of the following substituted cyclohexanes has the most negative value of ΔG° for ring flipping from the conformation in which the substituent is axial to the one where it is equatorial?
 - a. methylcyclohexane
 - b. chlorocyclohexane
 - c. isopropylcyclohexane
 - d. ethynylcyclohexane

ANSWER:

POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

- 60. Which of the following substituted cyclohexanes has the most negative value of ΔG° for ring flipping from the conformation in which the substituent is axial to the one where it is equatorial?
 - a. fluorocyclohexane
 - b. methylcyclohexane
 - c. ethylcyclohexane
 - d. tert-butylcyclohexane

ANSWER: d
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

61. The Newman projection of the gauche conformation of 1,2-dichloroethane is shown below.



a. True

b. False

ANSWER: True POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/10/2017 1:33 AM

- 62. The most stable conformation of an alkane occurs when carbon-carbon bonds are staggered and bulky groups are anti.
 - a. True
 - b. False

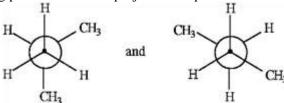
ANSWER: True POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

63. The following pairs of Newman projections represent the same compound but in differing conformations.



a. True

b. False

ANSWER: False POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/10/2017 1:38 AM

- 64. There are four constitutional isomers for the molecular formula C₆H₁₄.
 - a. True
 - b. False

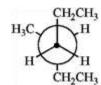
ANSWER: False POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

65. The following Newman projection represents 2-methylhexane.



a. True

b. False

ANSWER: False

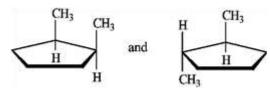
POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/10/2017 1:37 AM

66. The following structures represent, from left to right, a cis and a trans isomer.



a. True

b. False

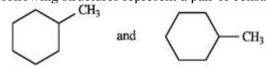
ANSWER: True POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

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67. The following structures represent a pair of constitutional isomers.



a. True

b. False

ANSWER: False

POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/10/2017 1:38 AM

68. 3-methylhexylcyclopentane represents a correct IUPAC name.

a. True

b. False

ANSWER: False

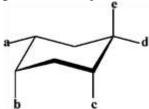
POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

69. In the following structure the positions labeled a and d are equatorial positions.



a. Trueb. False

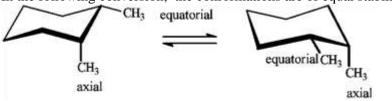
ANSWER: True POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/13/2017 2:28 AM

70. In the following conversion, the conformations are of equal stability.



a. True

b. False

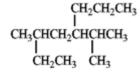
ANSWER: True POINTS: 1

QUESTION TYPE: True / False

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/10/2017 4:39 AM

71. The correct IUPAC name for the following compound is____



ANSWER: 3-methyl-5-(1-methylethyl)octane

3-methyl-5-isopropyloctane

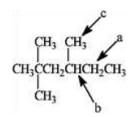
POINTS: 1

QUESTION TYPE: Completion

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/13/2017 2:29 AM

72. The tertiary carbon atom in the following structure is indicated by the letter_____.



ANSWER: b

В

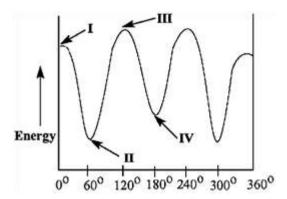
POINTS: 1

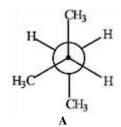
QUESTION TYPE: Completion

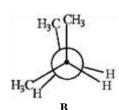
HAS VARIABLES: False

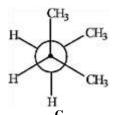
DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/13/2017 2:30 AM

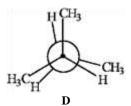
Match the Newman projection for the conformation of 2-methylbutane to the indicated position on the potential energy diagram.











73. Conformation A is represented by Roman numeral _____.

ANSWER: II POINTS: 1

QUESTION TYPE: Completion

HAS VARIABLES: False

PREFACE NAME: PE Diagram

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

74. Conformation B is represented by Roman numeral _____.

ANSWER: III POINTS: 1

QUESTION TYPE: Completion

HAS VARIABLES: False

PREFACE NAME: PE Diagram

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

75. Conformation C is represented by Roman numeral _____.

ANSWER: IV POINTS: 1

QUESTION TYPE: Completion

HAS VARIABLES: False

PREFACE NAME: PE Diagram

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

76. Conformation D is represented by Roman numeral _____.

ANSWER: I POINTS: 1

QUESTION TYPE: Completion

HAS VARIABLES: False

PREFACE NAME: PE Diagram

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77. What is the IUPAC name of the following compound?

ANSWER: 4-ethyl-2-methylhexane

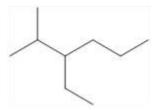
POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

78. What is the IUPAC name of the following compound?



ANSWER: 3-ethyl-2-methylhexane

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

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79. What is the IUPAC name of the following compound?



ANSWER: 2,4,4-trimethylhexane

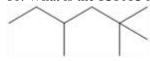
POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

80. What is the IUPAC name of the following compound?



ANSWER: 2,2,4-trimethylhexane

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

81. What is the IUPAC name of the following compound?



ANSWER: bicyclo[4.2.0]octane

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

82. What is the IUPAC name of the following compound?



ANSWER: bicyclo[2.2.1]heptane

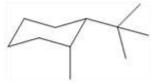
POINTS:

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

83. What is the IUPAC name of the following compound?



ANSWER: cis-1-tert-butyl-2-methylcyclohexane

[ignoring absolute stereochemistry in Chap 2]

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

84. What is the IUPAC name of the following compound?



ANSWER: cis-1-isopropyl-3-methylcyclohexane

[ignoring absolute stereochemistry in Chap 2]

POINTS:

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

85. How many hydrogen atoms are there in decane?

ANSWER: 22
POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

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86. How many hydrogen atoms are there in octane?

ANSWER: 18
POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

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87. What are the common and IUPAC names of the following compound?



ANSWER: common: neopentane

IUPAC: 2,2-dimethylpropane

POINTS:

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

88. What are the common and IUPAC names of the following compound?



ANSWER: common: isopentane

IUPAC: 2-dimethylbutane

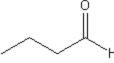
POINTS:

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

89. What is the IUPAC name of the following compound?



ANSWER: butanal

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

90. What is the IUPAC name of the following compound?



ANSWER: butanone

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

91. Provide a line-bond structure of hexanoic acid.

ANSWER:

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

92. Provide a line-bond structure of pentanal.

ANSWER:

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

93. Provide a Newman projection of the most stable conformation of 2-methylpentane, $(CH_3)_2CHCH_2CH_3$, looking along the C2-C3 bond

ANSWER:

POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/10/2017 5:50 AM

94. Provide a Newman projection of the most stable conformation of 3-methylpentane, CH₃CH₂CH(CH₃)CH₂CH₃ looking along the C2-C3 bond.

ANSWER:

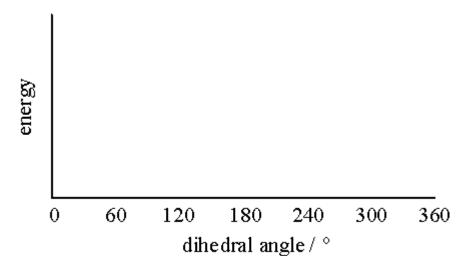
POINTS:

QUESTION TYPE: Subjective Short Answer

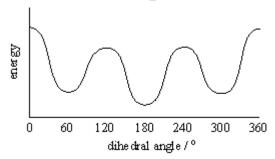
HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

95. Provide a neatly drawn plot of energy versus dihedral angle for rotation around the C2-C3 bond of butane.



ANSWER:



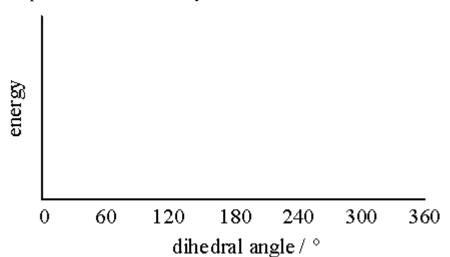
POINTS:

QUESTION TYPE: Subjective Short Answer

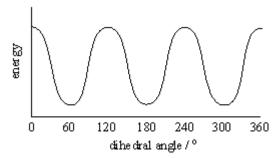
HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM *DATE MODIFIED:* 3/12/2015 3:24 PM

96. Provide a neatly drawn plot of energy versus dihedral angle for rotation around the C-C bond of ethane.



ANSWER:



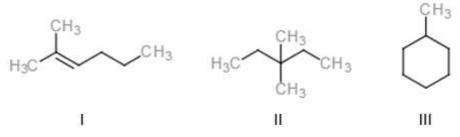
POINTS: 1

QUESTION TYPE: Subjective Short Answer

HAS VARIABLES: False

DATE CREATED: 3/12/2015 3:24 PM DATE MODIFIED: 3/12/2015 3:24 PM

97. Identify the constitutional isomer with the molecular formula C₇H₁₆.



a. Only I

b. Only I and II

c. Only II

d. I, II, and III

ANSWER: c
POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 4/11/2017 7:38 AM

DATE MODIFIED: 4/13/2017 1:02 AM

98. The molecular formulas of neopentane and isobutane are _____ and _____, respectively.

a. C₅H₁₂; C₄H₈

b. C₅H₁₀; C₄H₁₀

c. C₅H₁₂; C₄H₁₁

d. C₅H₁₂; C₄H₁₀

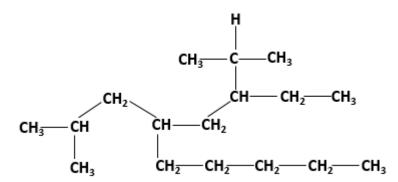
ANSWER: d POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 4/11/2017 7:54 AM DATE MODIFIED: 4/11/2017 7:55 AM

99. The total numbers of 2° and 3° carbon atoms in the following structure are _____ and _____, respectively.



a. 3; 7

b. 7; 4

c. 4; 6

d. 6; 4

ANSWER: b POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 4/11/2017 7:57 AM DATE MODIFIED: 4/11/2017 8:00 AM

100. What is "X" in the following reaction?

d. H₃C-CH₃

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 4/11/2017 8:01 AM DATE MODIFIED: 4/11/2017 8:09 AM

101. What is "X" in the following reaction?

$$H_3C$$
 CH_3
 $Catalyst$
 $-H_2$
 $Catalyst$
 $-3H_2$

a. (

b. CH₃

c. H₃C CH₃

d. H₂C

ANSWER: a POINTS: 1

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 4/11/2017 8:04 AM DATE MODIFIED: 4/11/2017 8:09 AM

102. What is "X" in the following reaction?

d. H_3C-CH_3

ANSWER: d

1

POINTS:

QUESTION TYPE: Multiple Choice

HAS VARIABLES: False

DATE CREATED: 4/11/2017 8:10 AM DATE MODIFIED: 4/19/2017 12:56 AM

Match the structures with their IUPAC names.

a. H₃C CH₃

 H_3C H_3C H_3C CH_3 CH_3

c. CH_3 CH_3 CH_3

QUESTION TYPE: Matching HAS VARIABLES: False

DATE CREATED: 4/11/2017 8:15 AM DATE MODIFIED: 4/11/2017 8:26 AM

103. 3-ethyl-2-methylhexane

ANSWER: a POINTS: 1

104. 4-ethyl-2,2-dimethylhexane

ANSWER: b
POINTS: 1

105. 4-ethyl-2-methylhexane

ANSWER: c
POINTS: 1